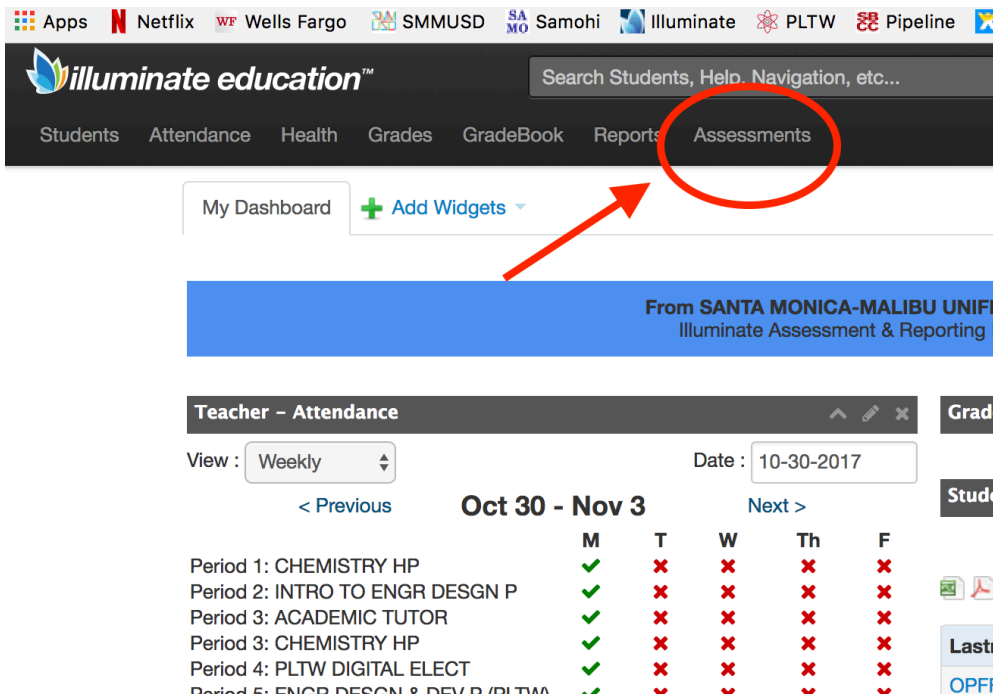
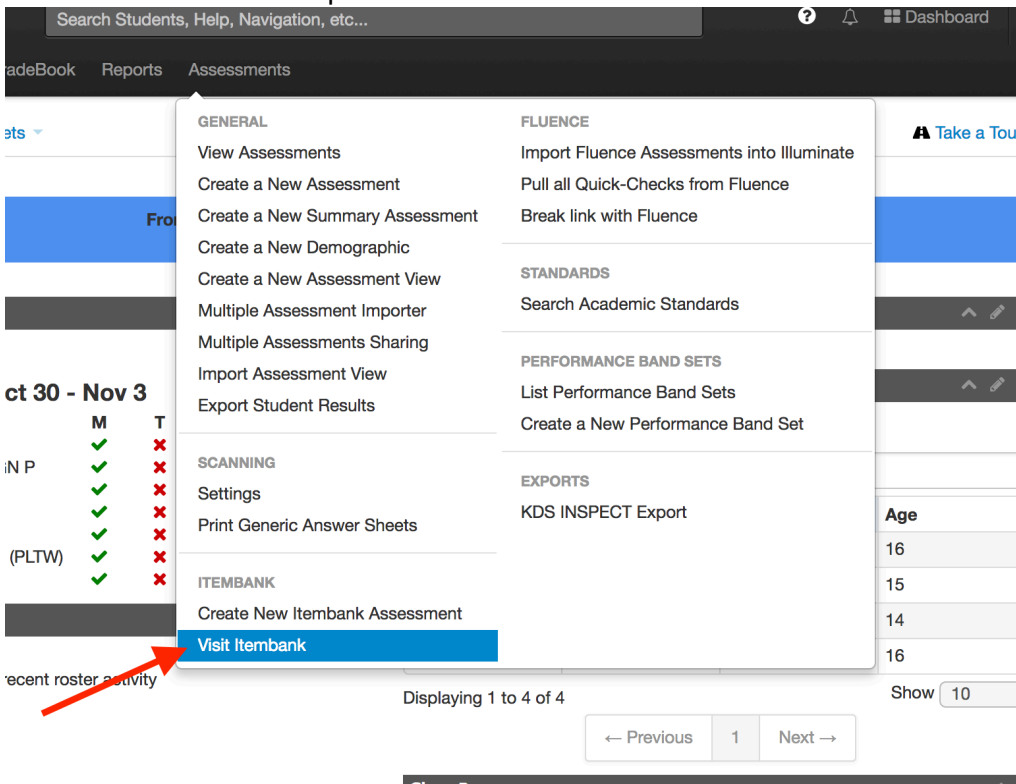


## How to Create Assessments Using the "Test Banks" in Illuminate:

1. Login to Illuminate
2. Click the Assessment Tab

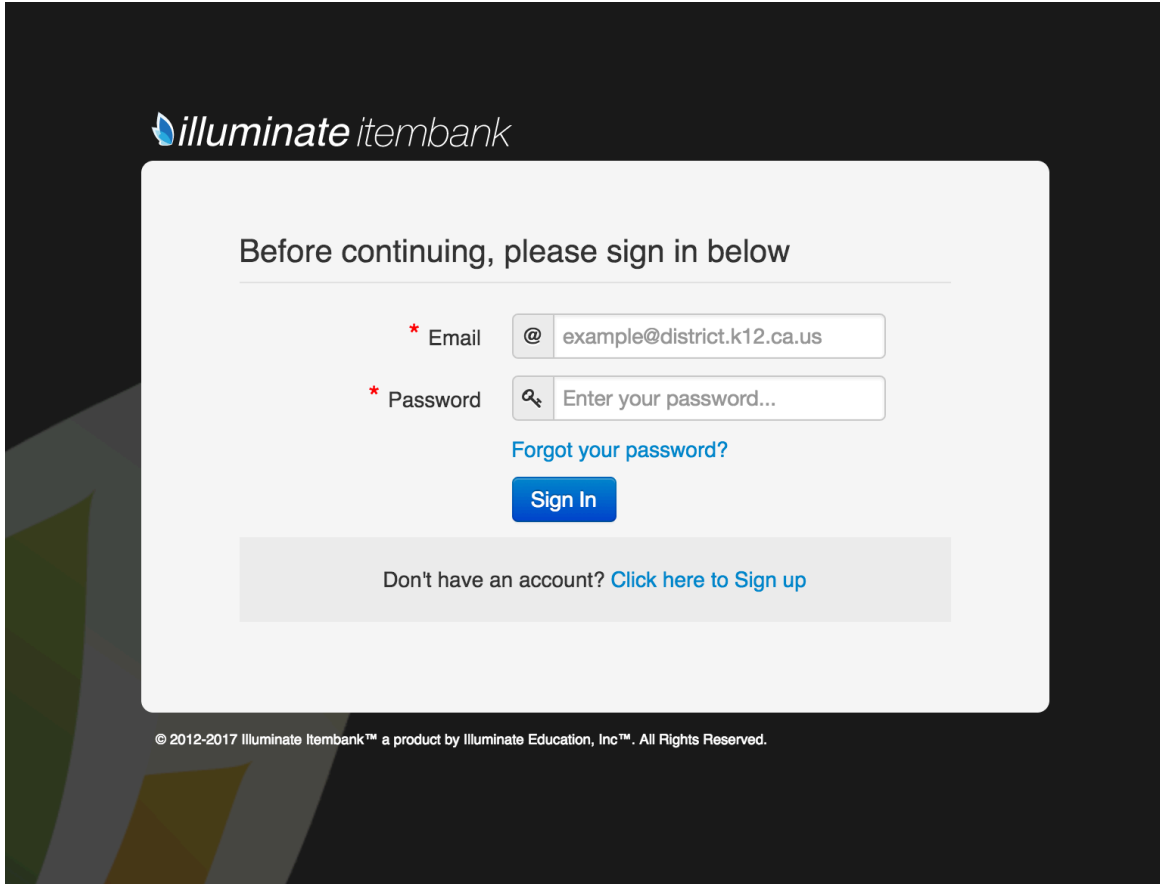


3. At the bottom of the dropdown menu select "Visit the Itembank"



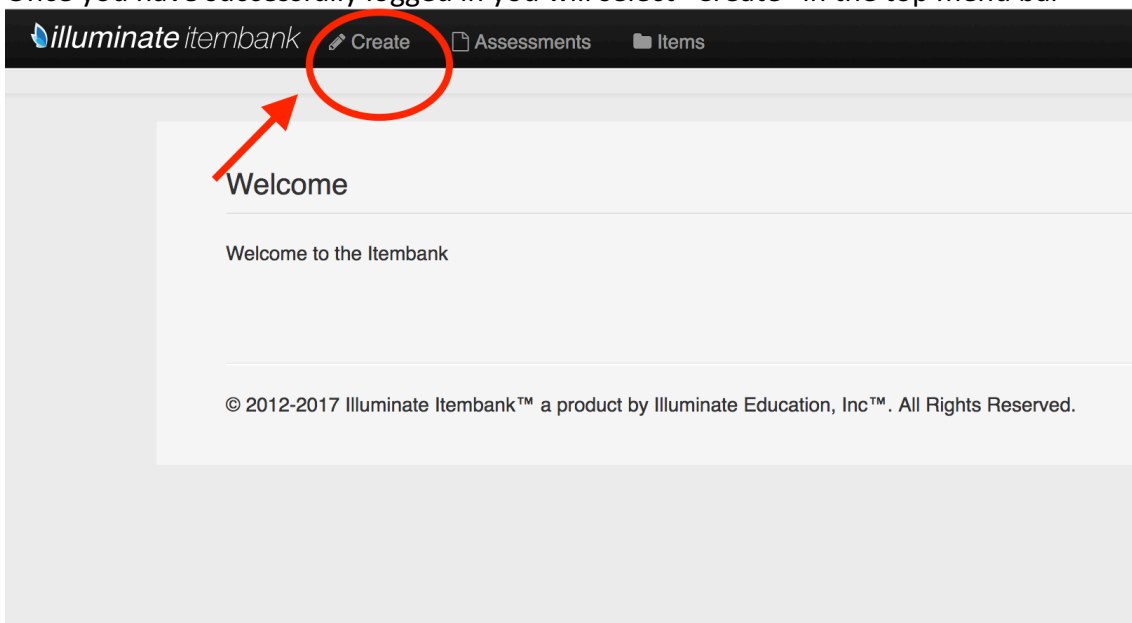
\*Note: If you have not used the Item bank feature before then you will need to login for the first time using your smmusd login and password at the following link

Item bank login link here: <https://itembank.illuminateed.com/auth/login>



The screenshot shows the login interface for Illuminate Itembank. At the top left is the logo "illuminate itembank". The main heading is "Before continuing, please sign in below". Below this are two input fields: "Email" with a red asterisk and a placeholder "@ example@district.k12.ca.us", and "Password" with a red asterisk and a placeholder "Enter your password...". A "Forgot your password?" link is positioned below the password field. A blue "Sign In" button is centered below the fields. At the bottom of the form area, there is a link: "Don't have an account? [Click here to Sign up](#)". At the very bottom of the page, there is a copyright notice: "© 2012-2017 Illuminate Itembank™ a product by Illuminate Education, Inc™. All Rights Reserved."

4. Once you have successfully logged in you will select "Create" in the top menu bar



5. Click "Assessment (Standard Mode)" to access standards from multiple published test banks and to add in your own questions if you have created your own Question Bank.

Hi there. What kind of content do you want to create?

- Assessment (Standard Mode)**  
Choose this to create your own test. You can select from any of the available public items, use any of your personal items, or use any from the available item banks.
- Assessment (Quick Mode)**  
This is the new approach to creating assessments. You will be able to select all the standards you want to test against. Furthermore, you will be able to set criteria for each selected standard. Questions will be auto generated based off the selected standards and set criteria.
- Item**  
Choose this to create your own test question. You can create questions just for you, or you can share them with other teachers.
- Passage**  
Choose this to create your own question Passage. Usually a Reading or Science Passage that precedes a group of questions on a test.

Next Cancel

6. Name your assessment and hit create

illuminate itebank Create Assessments Items BREANNA SNYDER

New Assessment Details Back to Assessments

Create Assessment Guide Me Create

\* Assessment Name: Sample Assessment

Assessment Description

Assessment Tags

Academic Year: -- Select Academic Year --

Grade Levels: Select Grade Levels

Subject: -- Select a Subject --

Create

- You will now be directed to a section that asks you which standards you will be addressing. This section allows you to access many different item banks so you may want to spend some time looking at the test banks that correspond to your subject matter.

**Standards**

What Standards are you Assessing?

-- Type -- -- All Subjects -- -- All Grade Levels --

Standard Keywords: NS, 1, inequalities, etc

**Recent:** SCI.9-12.CH.1.g SCI.9-12.CH.1.d

**Selected Standards**

No Standards Linked

**Materials**

Add Material

Title
No Materials Available

As a reference the top menu bar will show you what you still need to do in your assessment. You will work from left to right in order to create your assessment using the Illuminate Item bank.

**illuminate itembank** Create Assessments Items BREANNA SNYDER

New Assessment Details Back to Assessments

**Create Assessment** Guide Me Create

\* Assessment Name: Sample Assessment

Assessment Description

**Assessment Tags**

Academic Year: -- Select Academic Year --

Grade Levels: Select Grade Levels

Subject: -- Select a Subject --

Create

## 8. Select your "Standard Type" and "Subject"

Standards

What Standards are you Assessing?

Advanced Placement Standards Science (2012) Advanced Placement Chemistry

Standard Keywords: NS.1.1, Inequalities, etc

Recent: SCI.9-12.CH.1.g SCI.9-12.CH.1.d

Expand All Collapse All

## 9. Once you have selected the standards that you want addressed, items will pop-up for you to select to create a test from.

▼ Advanced Placement Standards

▼ Science

▼ Advanced Placement Chemistry

- ▶ **SCI.9-12.1 - ( 21 Items )**  
The chemical elements are fundamental building materials of matter, and all matter can be understood in terms of arrangements of atoms. These atoms retain their identity in chemical reactions.
- ▶ **SCI.9-12.2 - ( 15 Items )**  
Chemical and physical properties of materials can be explained by the structure and the arrangement of atoms, ions, or molecules and the forces between them.
- ▶ **SCI.9-12.3 - ( 3 Items )**  
Changes in matter involve the rearrangement and/or reorganization of atoms and/or the transfer of electrons.
- ▶ **SCI.9-12.4 - ( 0 Items )**  
Rates of chemical reactions are determined by details of the molecular collisions.
- ▶ **SCI.9-12.5 - ( 1 Item )**  
The laws of thermodynamics describe the essential role of energy and explain and predict the direction of changes in matter.
- ▶ **SCI.9-12.6 - ( 0 Items )**  
Any bond or intermolecular attraction that can be formed can be broken. These two processes are in a dynamic competition, sensitive to initial conditions and external perturbations.

## 10. Select items that you want to add to your test by clicking the link.

▼ Advanced Placement Standards

▼ Science

▼ Advanced Placement Chemistry

- ▶ **SCI.9-12.1 - ( 21 Items )** Selected  
The chemical elements are fundamental building materials of matter, and all matter can be understood in terms of arrangements of atoms. These atoms retain their identity in chemical reactions.
- ▶ **SCI.9-12.2 - ( 15 Items )**  
Chemical and physical properties of materials can be explained by the structure and the arrangement of atoms, ions, or molecules and the forces between them.
- ▶ **SCI.9-12.3 - ( 3 Items )**  
Changes in matter involve the rearrangement and/or reorganization of atoms and/or the transfer of electrons.
- ▶ **SCI.9-12.4 - ( 0 Items )**  
Rates of chemical reactions are determined by details of the molecular collisions.
- ▶ **SCI.9-12.5 - ( 1 Item )** Selected  
The laws of thermodynamics describe the essential role of energy and explain and predict the direction of changes in matter.
- ▶ **SCI.9-12.6 - ( 0 Items )**  
Any bond or intermolecular attraction that can be formed can be broken. These two processes are in a dynamic competition, sensitive to initial conditions and external perturbations.

11. If you click on the “Expand All” It will show you more specific questions in the standard that you have selected. This function will allow you to be more specific when choosing the questions to be in your assessment.

▼ **Advanced Placement Standards**

- ▼ **Science**
  - ▼ **Advanced Placement Chemistry**
    - ▼ **SCI.9-12.1 - ( 21 Items )** Selected
      - The chemical elements are fundamental building materials of matter, and all matter can be understood in terms of arrangements of atoms. These atoms retain their identity in chemical reactions.
      - ▼ **SCI.9-12.1.A - ( 3 Items )**
        - All matter is made of atoms. There are a limited number of types of atoms; these are the elements.
        - ▼ **SCI.9-12.1.A.1 - ( 1 Item )**
          - Molecules are composed of specific combinations of atoms; different molecules are composed of combinations of different elements and of combinations of the same elements in differing amounts and proportions.
          - ▼ **SCI.9-12.1.A.1.a - ( 0 Items )**
            - The average mass of any large number of atoms of a given element is always the same for a given element.
            - ▼ **SCI.9-12.1.A.1.b - ( 0 Items )**
              - A pure sample contains particles (or units) of one specific atom or molecule; a mixture contains particles (or units) of more than one specific atom or molecule.

12. Once you have selected all of the “items” that you want to be on your assessment select “continue”

**SCI.9-12.2.A.3.g - ( 0 Items )**  
The formation of a solution may be an exothermic or endothermic process, depending on the relative strengths of intermolecular/interparticle interactions before and after the dissolution process.

**SCI.9-12.2.A.3.h - ( 1 Item )**  
Generally, when ionic compounds are dissolved in water, the component ions are separated and dispersed. The presence of ions in a solution can be detected by use of conductivity measurements.

**SCI.9-12.2.A.3.i - ( 0 Items )**  
Solution composition can be expressed in a variety of ways; molarity is the most common method used in the laboratory. Molarity is defined as the number of moles of solute per liter of solution.

**SCI.9-12.2.A.3.j - ( 0 Items )**  
Understanding how to prepare solutions of specified molarity through direct mixing of the components, through use of volumetric glassware, and by dilution of a solution of known molarity with additional solvent is important for performing laboratory work in chemistry.

**SCI.9-12.LO 2.7 - ( 0 Items )**  
The student is able to explain how solutes can be separated by chromatography based on intermolecular interactions.

**SCI.9-12.LO 2.8 - ( 0 Items )**  
The student can draw and/or interpret representations of solutions that show the interactions between the solute and solvent.

**Summary:**

0	2	0
Items	Standards	Passages

**Passages**

**Standards**

- SCI.9-12.1 (0)
- SCI.9-12.5 (0)

[Question Groups Overview](#)

Created By (BREANNA SNYDER)  
Created (10-30-17 6 PM)  
Updated (10-30-17 6 PM)

13. The following screen will pop-up which will allow you to add additional filters to access other question banks.

**Search Filters** ⓘ

Selected Standards [SCI.9-12.1](#) [SCI.9-12.5](#)

Item Banks [My Items / Passages, Inspect, Santa Monica Malibu Unified School District](#)

Authors [Select Authors](#)

Item Types [Select Item Types](#)

Item Keywords [Enter Item Keywords](#)

Passages [With or Without Passages](#)

Passage Title [Enter Passage Title](#)

Attributes [Item \(Language\(English\)\)](#)

Item / Passage Identifiers [Enter Identifiers](#)

[Reset](#) [Search](#)

If you select “Item Banks” then can choose which banks you want to access. The “public” item bank may not have the most valid question types because any person can write these questions and submit to the bank.

**Select Item Banks** X

Itembanks  My Items / Passages

Public

CoreSpring Open Collection

EduThink

Inspect

Santa Monica Malibu Unified School District

[Update Filters](#) [Cancel](#)

If you select “Item Types” then the following screen will appear and you can select the type of questions that you want to add to your assessment.

### Select Item Types x

<input type="checkbox"/>	<b>Item Type</b>
<input type="checkbox"/>	Multiple Choice / Selected Response (one correct)
<input type="checkbox"/>	Non-Traditional Selected Response (select all, T/F, Y/N, C/I)
<input type="checkbox"/>	Evidence Based Selected Response
<input type="checkbox"/>	Highlight - Selectable Text
<input type="checkbox"/>	Highlight - Shading
<input type="checkbox"/>	Highlight - Hotspot
<input type="checkbox"/>	Select Dropdown
<input type="checkbox"/>	Constructed Response
<input type="checkbox"/>	Explicit Constructed Response

Update Filters Cancel

14. Once you have narrowed your selections a list of potential questions will appear at the bottom of your page. Select the questions that you want to add and hit “continue”

Search Results

Found 0 Passage(s), 37 Item(s)

⏪ ⏩ 1 2 3 4 ⏪ ⏩

**Question** MC SCI.9-12.1.C

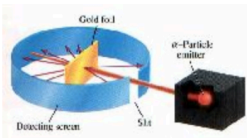
By knowing the number of electrons in a neutral atom, you should also be able to determine

- A the atomic number of the neutral atom
- B two of these
- C the number of neutrons in the neutral atom
- D the number of protons in the neutral atom
- E the mass of the neutral atom

[Click to show Rationale & Item Info](#) Add

**Question** MC SCI.9-12.1.D

The apparatus pictured here helped to determine:



- A the existence of electrons



15. Once you have selected questions you will be able to “review” your selections and make changes. If you are satisfied with the questions that you have selected, then hit “continue”.

illuminate itembank Create Assessments Items

Sample Assessment Details Add Items **Review** Overview Booklet Publish

### Questions

**1** **Question MC** **Item In Use** **SCI.9-12.1, SCI.9-12.1.C**

By knowing the number of electrons in a neutral atom, you should also be able to determine

- A the atomic number of the neutral atom
- B two of these
- C the number of neutrons in the neutral atom
- D the number of protons in the neutral atom
- E the mass of the neutral atom

16. In the overview section you are able to add a section to your test if you desire. Hit “continue” if you do not want to add sections.

Sample Assessment Details Add Items Review **Overview** Booklet Publish

### Questions Overview

**Section 1** **2 items**

<b>1</b> <b>MC</b> By knowing the number of electrons in	SCI.9-12.1, SCI.9-12.1.C
<b>2</b> <b>MC</b> The apparatus pictured here helped to	SCI.9-12.1, SCI.9-12.1.D

**Add Section**

**Add Section**

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17. The following screen will appear which will allow you to customize how the assessment is printed in handout form for the students. Once you have selected your format hit “generate” and select a location to save your assessment to your computer.

## Print Options Generate Saved

**\* Font Face**

Helvetica (Sans-Serif)

Times (Serif)

DejaVu (Serif)

DejaVu (Sans-Serif)

OpenDyslexic (Sans-Serif)

**\* Font Size**

Tiny - 8pt

Small - 10pt

Med - 12pt

L - 14pt

XL - 16pt

**\* Question Spacing**

Minimal - 1x

Medium - 1.5x

Large - 2x

XL - 3x

XXL - 4x

XXXL - 10x

**\* Choice Spacing**

Minimal - 1x

Medium - 1.5x

Large - 2x

XL - 3x

XXL - 4x

XXXL - 10x

**\* Constructed Response Space Formatting**

Lined

Blank Space

**\* Constructed Response Spacing**

None

1 Line

3 Lines

5 Lines

10 Lines

20 Lines

**\* Teacher Edition**

Yes

No

**\* Teacher Edition Options**

Shows on Teacher Edition only

Rationale

Attributes

**\* Cover Page**

Include

Do not include

18. Once you have saved your assessment to your computer hit “continue” and “publish your assessment”

19. After you publish your assessment you can access this assessment in your normal illuminate account.

Students Attendance Health Grades GradeBook Reports Assessments

View Assessments

If you don't care for the new version of the assessment list, you can return to the [previous version](#). You are currently viewing the new assessment list.

New

More Filters ▾
Search

Displaying 1 to 50 of 443

ID	Type	Title	Owner	Date Created	Last Accessed	Actions
<input type="checkbox"/>	Itembank	Sample Assessment	SNYDER, BREANNA	October 31, 2017	October 31, 2017	Action ▾

- ☰ All Assessments
- 👤 My Assessments
- ★ Favorites