

**Neutralizations Reactions***Practice Sheet #41*

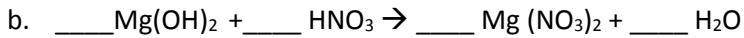
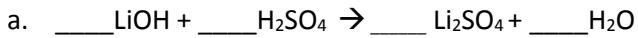
1. Classify each compound as an acid or base and name the compound.

- a. RbOH
- b. HF
- c. H<sub>3</sub>AsO<sub>3</sub>
- d. HOCl
- e. Al(OH)<sub>3</sub>

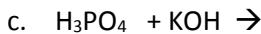
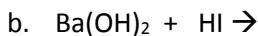
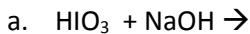
2. Write the formula for each compound.

- a. Hydrobromic acid
- b. Strontium hydroxide
- c. Carbonous acid
- d. Acetic acid
- e. Bromic acid
- f. Cesium hydroxide

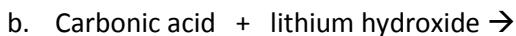
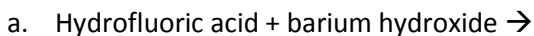
3. Balance each neutralization reaction.



4. Complete and balance each neutralization reaction



5. Write the formula for each compound and complete and balance the neutralization reaction.



Name: \_\_\_\_\_ Per \_\_\_\_\_

6.